

# 2017 Consumer Confidence Report

Water System Name: Yosemite Spring Park Utility Co. Report Date: June 18, 2018

*We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 - December 31, 2017 and may include earlier monitoring data.*

**Este informe contiene información muy importante sobre su agua potable. Tradúzcalo ó hable con alguien que lo entienda bien.**

Type of water source(s) in use: 100% of our water is produced from deep hard rock wells.

Name & general location of source(s): Wells: 1A, 31A, 36A, 37A, 39A, 40A, 42A, 45A, 46A, 47A, & 49A.  
All of our wells are located within the boundaries of the Yosemite Lakes Park Subdivision.

Drinking Water Source Assessment information: The following sources, Wells 1A, 31A, 36A, 37A, 39A, 40A, 42A, 45A, 46A, 47A and 49A are considered most vulnerable to the following activities not associated with any detected contaminants: Septic Systems – Low Density (<1/acre).

Time and place of regularly scheduled board meetings for public participation: Our regularly scheduled Board Meetings are held at 6:00pm on the 3rd Tuesday every month in the Yosemite Lakes Clubhouse located at 30250 Yosemite Springs Parkway.

For more information, contact: Kenneth Harrington Phone: (559) 658-7451

## TERMS USED IN THIS REPORT

**Maximum Contaminant Level (MCL):** The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

**Maximum Contaminant Level Goal (MCLG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

**Public Health Goal (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Maximum Residual Disinfectant Level (MRDL):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**Maximum Residual Disinfectant Level Goal (MRDLG):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standards (PDWS):** MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Secondary Drinking Water Standards (SDWS):** MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

**Treatment Technique (TT):** A required process intended to reduce the level of a contaminant in drinking water.

**Regulatory Action Level (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Variations and Exemptions:** State Board permission to exceed an MCL or not comply with a treatment technique under certain conditions.

**Level 1 Assessment:** A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

**Level 2 Assessment:** A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

**ND:** not detectable at testing limit

**ppm:** parts per million or milligrams per liter (mg/L)

**ppb:** parts per billion or micrograms per liter (µg/L)

**ppt:** parts per trillion or nanograms per liter (ng/L)

**ppq:** parts per quadrillion or picogram per liter (pg/L)

**pCi/L:** picocuries per liter (a measure of radiation)

**The sources of drinking water** (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

**Contaminants that may be present in source water include:**

- *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- *Pesticides and herbicides*, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- *Organic chemical contaminants*, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- *Radioactive contaminants*, that can be naturally-occurring or be the result of oil and gas production and mining activities.

**In order to ensure that tap water is safe to drink**, the U.S. EPA and the State Water Resources Control Board (State Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. State Board regulations also establish limits for contaminants in bottled water that provide the same protection for public health.

**Tables 1, 2, 3, 4, 5, and 6 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent.** The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

**TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA**

Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of Months in Violation	MCL	MCLG	Typical Source of Bacteria
Total Coliform Bacteria (state Total Coliform Rule)	(In a mo.) 0	0	1 positive monthly sample	0	Naturally present in the environment
Fecal Coliform or <i>E. coli</i> (state Total Coliform Rule)	(In the year) 0	0	A routine sample and a repeat sample are total coliform positive, and one of these is also fecal coliform or <i>E. coli</i> positive	0	Human and animal fecal waste

**TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER**

Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of Samples Collected	90 <sup>th</sup> Percentile Level Detected	No. Sites Exceeding AL	AL	PHG	No. of Schools Requesting Lead Sampling	Typical Source of Contaminant
Lead (ppb)	9/27/16 & 9/28/16	20	9.2	0	15	0.2	Not applicable	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits
Copper (ppm)	9/27/16 & 9/28/16	20	016	0	1.3	0.3	Not applicable	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives

**TABLE 3 – SAMPLING RESULTS FOR SODIUM AND HARDNESS**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Sodium (ppm)	12-2014 / 12-2017	27.0 AVG.	20 to 43	none	none	Salt present in the water and is generally naturally occurring
Hardness (ppm)	12-2014 / 12-2017	134.69 AVG.	66 to 240	none	none	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring

**TABLE 4 – DETECTION OF CONTAMINANTS WITH A PRIMARY DRINKING WATER STANDARD**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant
Arsenic Level (ppb)	12-2014 / 12-2017	2.98 AVG.	1.0 to 7.0	10	0.004	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes.
Barium (ppm)	12-2014 / 12-2017	.0026 AVG.	ND to 0.014	1	2	Discharges of oil drilling wastes and from metal refineries; erosion of natural deposits.
<i>Fluoride Level (ppm)</i>	<i>12-2014 / 12-2017</i>	<i>.45 AVG.</i>	<i>ND to 2.4*</i>	<i>2.0</i>	<i>1</i>	<i>Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories.</i>
Nickel (ppb)	12-2014 / 12-2017	0.30 AVG.	ND to 2.1	100	12	Erosion of natural deposits; discharge from metal factories.
Total Trihalomethanes (ppb)	7-12-17	35		80	N/A	By-product of drinking water disinfection.
Haloacetic Acids (ppb)	7-12-17	14		60	N/A	By-product of drinking water disinfection.
Gross Alpha (pCi/L)	6-2008 / 10-2015	2.96 AVG.	ND to 11.83	15	0	Erosion of natural deposits.
Uranium (pCi/L)	10-2012 / 10-2015	3.15 AVG.	ND to 17	20	.43	Erosion of natural deposits.
Radium 228 (pCi/L)	6-2008 / 6-2010	0.31 AVG.	ND to 1.56	5	0.019	Erosion of natural deposits.
Chlorine (ppm)	1-2017 / 12-2017	.98 AVG.	0.25 to 1.80	4.0	4	Drinking water disinfectant added for treatment.

**TABLE 5 – DETECTION OF CONTAMINANTS WITH A SECONDARY DRINKING WATER STANDARD**

Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant
Color (units)	12-2014 / 12-2017	4.23 AVG.	ND to 15	15	N/A	Naturally occurring organic materials.
Odor Threshold (units)	12-2014 / 12-2017	1.35 AVG.	1 to 3	3	N/A	Naturally occurring organic materials.
<i>Iron (ppb)</i>	12-2014 / 12-2017	<i>508</i> AVG.	<i>ND to 1700*</i>	<i>300</i>	<i>N/A</i>	<i>Leaching from natural deposits; industrial wastes.</i>
<i>Manganese (ppb)</i>	12-2014 / 12-2017	<i>1590.92</i> AVG.	<i>ND to 18000*</i>	<i>50</i>	<i>N/A</i>	<i>Leaching from natural deposits.</i>
<i>Turbidity (units)</i>	12-2014 / 12-2017	<i>3.09</i> AVG.	<i>ND to 12*</i>	<i>5</i>	<i>N/A</i>	<i>Soil Runoff.</i>
Zinc (ppm)	12-2014 / 12-2017	.10 AVG.	ND to 1.1	5	N/A	Runoff/leaching from natural deposits; industrial wastes.
Total Dissolved Solids [TDS] (ppm)	12-2014 / 12-2017	256.2 AVG.	190 to 400	1000	N/A	Runoff/leaching from natural deposits.
Specific Conductance (umho/cm)	12-2014 / 12-2017	403.8 AVG.	300 to 630	1600	N/A	Substances that form ions when in water; seawater influence.
Chloride (ppm)	12-2014 / 12-2017	13.69 AVG.	4.3 to 25	500	N/A	Runoff/leaching from natural deposits; seawater influence.
Sulfate (ppm)	12-2014 / 12-2017	33.18 AVG.	4.3 to 120	500	N/A	Runoff/leaching from natural deposits; industrial wastes.

\*Any violation of an MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

### Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

**Lead-Specific Language for Community Water Systems:** If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. The Yosemite Spring Park Utility Company is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4701) or at <http://www.epa.gov/lead>.

### Summary Information for Secondary Contaminants in Excess of the MCL

Iron and manganese were found at levels that exceed the secondary MCL of 300 ug/L and 50 ug/L respectfully. The wells that test the highest for iron are only used based on customer demand. The iron and manganese MCLs were set to protect you against unpleasant aesthetic effects (e.g., color, taste, and odor) and the staining of plumbing fixtures (e.g., tubs and sinks) and clothing while washing. The high iron and manganese levels are due to leaching of natural deposits in the Earth. The Yosemite Spring Park Utility Company water system operates under a waiver for the State Water Resources Control Board (SWRCB) due to these minerals. Your Utility understands that water, especially chlorinated water, containing high levels of iron and manganese causes discoloration that can be very frustrating to the consumer. Because of this we have taken extra ordinary steps to control the adverse effects that result from these minerals. In 1996 we began pioneering a process to reduce the affects that these minerals cause. After a three year pilot study we received authorization from SWRCB to provide a specialized treatment to control the oxidation process of these minerals that causes the discoloration. While this process is not 100% effective, it does drastically reduce the number of occurrences of discolored water.

Three wells exceeded the MCL for Turbidity (Wells 45A and 49A). This turbidity is controlled through our sequestering treatment. Since these constituents are classified as secondary contaminants, there is no potential health effects language to provide to you. The problem associated with these contaminants affect the aesthetic properties of the water such as odor and color. There are no precautions that you as our customers need to take at this time.

Additionally, these contaminates are mitigated in the finished water through our use of the above mentioned specialized treatment method.

### For Water Systems Providing Groundwater as a Source of Drinking Water

**TABLE 7 – SAMPLING RESULTS SHOWING  
FECAL INDICATOR-POSITIVE GROUNDWATER SOURCE SAMPLES**

<b>Microbiological Contaminants</b> (complete if fecal-indicator detected)	<b>Total No. of Detections</b>	<b>Sample Dates</b>	<b>MCL [MRDL]</b>	<b>PHG (MCLG) [MRDLG]</b>	<b>Typical Source of Contaminant</b>
<i>E. coli</i>	(In the year) 0	Jan. 2017 thru Dec. 2017	0	(0)	Human and animal fecal waste
Enterococci	(In the year) 0	Jan. 2017 thru Dec. 2017	TT	n/a	Human and animal fecal waste
Coliphage	(In the year) 0	Jan. 2017 thru Dec. 2017	TT	n/a	Human and animal fecal waste